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Article

Spatio-temporal Investigations of the Incomplete Spin Transition in a Single Crystal of [Fe(2-pytrz)₂{Pt(CN)₄}]·3H₂O: Experiment and Theory

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Abstract: Optical microscopy technique is used to investigate the thermal and the spatio-temporal properties of the spin-crossover single crystal [Fe(2-pytrz)₂{Pt(CN)₄}]·3H₂O, which exhibits a first-order spin transition from a full high-spin (HS) state at high temperature to an intermediate, high-spin low-spin (HS-LS) state, below 153 K, where only one of the two crystallographic Fe(II) centers switches from the HS to HS-LS state. In comparison with crystals undergoing a complete spin transition, the present transformation involves smaller volume changes at the transition, which helps to preserving the crystal's integrity. By analyzing the spatio-temporal properties of this spin transition, we evidenced a direct correlation between the orientation and shape of HS/HS-LS domain wall with the crystal's shape. Thanks to the small volume change accompanying this spin transition, the analysis of the experimental data by an anisotropic reaction-diffusion model becomes very relevant and leads to an excellent agreement with the experimental observations.

Keywords: spin-crossover; optical microscopy; reaction diffusion

1. Introduction

Because of the growing societal requirement of information processing and big data storage, the design of devices with reversible and high density storage capacities as well as fast responses becomes mandatory. Multi-functional molecular materials with intrinsic physical properties at the molecular scale constitute serious candidates for their use as nano-memories, nano-switches or nano-probes due to many advantages, like their easy processability and their low cost. In this context, spin crossover (SCO) materials, based on iron (II) complexes, have been recognized as prime candidates, in particular due to their variety of nano-structuration possibilities [1–5] allowing their integration into devices for a various set of currently considered applications, e.g., as display and memory devices [6-9], sensing of temperature [10-12], probes of contact pressure or shocks [13], as well as actuators [14–16]. From the thermodynamic point of view, the SCO phenomenon is an entropic-driven mechanism involving the switching between two different electronic states, namely a diamagnetic low-spin (LS) state and a paramagnetic high-spin (HS) state, as a response to the application of an external stimulus. In the case of d⁶ Fe(II) based SCO compounds, the electronic configuration of the metal ion in the HS state, which generally emerges at high temperature, is (t_2^4, e_2^2) , where the valence electrons fill all electronic orbitals according to the Hund rule leading to the total spin momentum S = 2, which makes the SCO systems paramagnetic in this state. In contrast, in the

LS state, which appears at low-temperature, the electronic configuration $\mathsf{t}_{2g}^6\mathsf{e}_g^0$ is stabilized, for which corresponds a spin momentum, S=0, leading to a diamagnetic LS state.

On the other hand, due to the antibonding nature of the e_g orbital, the Fe-ligand distances increase by $\sim 10\%$ in the HS state in comparison with those of LS state, which induces an expansion of the unit cell volume in the HS state [17]. This molecular volume expansion and contraction in the HS and LS states, respectively, is delocalized in the lattice over several unit cells and is at the origin of the cooperative effects observed in these molecular solids. Indeed, as a consequence of these long-range interactions between molecules, insured by the lattice phonons, SCO solids can display first-order phase transitions accompanied with thermal hysteresis loops [6,18,19], allowing them to show macroscopic bistability, and then to switch collectively between the LS and the HS states. In addition, the spin transition can be triggered by various external parameters, such as pressure, light, temperature, electric and magnetic fields, [6,20,21], thus increasing their potential application in many areas.

From the microscopic point of view, the spin transition phenomenon has been considered as a vibronic problem [22,23] in which the electronic and vibrational structures of the molecule are strongly coupled, to the extent that in some cases, the Born-Oppenheimer approximation is not valid [24,25]. The microscopic changes of the magnetoelastic properties of the SCO solids at the transition, accompanied with large volume changes resulting from the constructive interferences of the molecular volume changes which deploy at long-range through elastic interactions, cause important variations in the physical properties of the SCO materials, which show thermo-chromic features at the transition as well as significant rigidity changes. As a direct consequence, an important panel of experimental techniques is utilized to study the SCO phenomenon, among them one can quote, the differential scanning calorimetry [26], magnetometry [27], X-ray diffraction [28], Mössbauer spectroscopy [29], diffuse reflectivity [30] as well as optical microscopy (OM) [31–38], which has been found very useful in the understanding of the non-equilibrium properties of the SCO materials. As a matter of fact, OM is now recognized as one of the major techniques allowing a direct spatiotemporal imaging of the thermally-induced spin crossover transition, with the respective spatial and temporal resolutions of $0.3~\mu m$ and 10~ms.

OM allows the study of the first-order phase transitions of SCO solids on a unique single crystal, which constitutes an undeniable progress towards a deep understanding of the fundamental aspects of this phenomenon. In addition, this technique provides valuable information not only on the electronic properties of the studied single crystals along the transition but also on the modification of their elastic properties, particularly when the spin transition is accompanied with a significant volume change. Among the large number of SCO compounds already investigated by OM [31–38], it is interesting to recall that the first quantitative experiments [35] were conducted on the fragile SCO single crystals of $[Fe(btr)_2(NCS)_2]H_2O$ (btr = 4,4'-bis-1,2,4-triazole), which easily lose watter molecules at the transition, causing their deterioration. To avoid such a disastrous effect, the single crystals were embedded in oil and the OM measurements have been performed by preventing any dehydration. Despite this, the first cooling process from HS to LS revealed the appearance of irreversible defects in the crystal caused by the mechanical stresses generated by the volume change at the spin transition, which hindered the complete analysis of the spatio-temporal properties along the two branches of the thermal hysteresis, although it was possible to evaluate the HS-LS interface velocity on cooling to \sim 2.3 μ m·s⁻¹. Later on, Chong et al. [36] investigated the spin transition in the hexagonally-shaped single crystals of [Fe(bbtr)₃](ClO₄)₂. Similarly to the previous study, the SCO transition was found to start at the corner of the hexagon, with a clear HS-LS interface which adapts its length and shape along its propagation process. Unfortunately, due to the large lattice parameter misfit between the LS and HS structures, the system does not succeed in finding a stress free interface, which causes irreversible crystal damages after the first cooling. It is worth mentioning that this enhanced brittleness of the crystals around the interface region which stores the excess of elastic energy, is quite well explained in the theoretical electro-elastic models [37,39–42] describing the elastic properties of the SCO solids. Finally, more recent

OM studies, performed on the resilient SCO single crystal [Fe(NCSe)(py)₂(μ -bpypz)] (py = pyridine and bpypz = 3,5-bis(2-pyridyl)-pyrazolate) [37,42–44], allowed for the first-time to monitor the motion of the HS-LS interface on both cooling and heating processes, whose corresponding velocities were estimated in the range 4–6 μ m·s⁻¹. In the latter case, due to the particularly regular shape of the crystals, made of rectangular platelets along the direction of the interface propagation, the shape of the HS-LS interface remained unchanged during its motion. Very recently [38], the SCO compound, $[Fe(2-pytrz)_2Pd(CN)_4]\cdot 3H_2O(2-pytrz = 4-(2-pyridyl)-1,2,4,4H-triazole)$, has been investigated by OM and showed a well defined interface during the spin transition. Thanks to the robust character of the investigated single crystals, the dynamical properties of the interface propagation were deeply studied, and their behavior were well reproduced by a reaction diffusion model. In the present work, we report about new OM investigations, realized on a high quality SCO complex of formula [Fe(2-pytrz)₂Pt(CN)₄]·3H₂O (1) for which we selected a reliable single crystal characterized by an irregular shape, in order to investigate the interplay between the mode of interface propagation and the crystal's shape. The expected synergetic effects are considered here as an evidence of the action of the long-range elastic interactions between the SCO molecules within the crystal, caused by the volume change at the transition. One of the consequences will be that mean-field based models become suitable for the theoretical description of the spatiotemporal properties of SCO materials, as will be demonstrated below.

The present manuscript is organized as follows: Section 2 summarizes the experimental findings on the thermo-induced spin transition of the $[Fe(2-pytrz)_2\{Pt(CN)_4\}]\cdot 3H_2O$ and presents the data of OM investigations which revealed the spatio-temporal character of the SCO transition. In Section 3, we introduce the theoretical spatio-temporal model, based on reaction diffusion description, allowing to faithfully describe the experimental data. Section 4 concludes and outlines some possible developments of this work.

2. Results and Discussion

The present work contains experimental investigations of optical microscopy performed at several temperatures in the thermal hysteresis region in order to observe the spatio-temporal features of the first-order spin transition of the title compound. To understand the physical mechanism of the interface propagation in this particular system, in which the spin transition is accompanied with a small volume change, we developed an appropriate modeling, based on a reaction diffusion description, allowing to reproduce the experimental spatiotemporal features of the current first-order SCO transition.

3. Experimental Investigations

3.1. Synthesis

Single crystals of $[Fe(2-pytrz)_2\{Pt(CN)_4\}]\cdot 3H_2O$ were obtained by slow diffusion, in a fine glass tube (3.0 mm diameter) of two different aqueous solutions: the first one was prepared by dissolving $K_2[Pt(CN)_4]\cdot xH_2O$ (37.7 mg, 0.1 mmol) in 10 mL; the second one was obtained by dissolving iron(II) perchlorate salt (25.5 mg, 0.1 mmol) in a solution (10 mL) of 4-(2-pyridyl)-1,2,4,4H-triazole (2-pytrz) (29.2 mg, 0.2 mmol), which became light yellow after a night. A volume of 1 mL of the $K_2[Pt(CN)_4]\cdot xH_2O$ solution was put in the fine glass tube and then similar volume (1 mL) was layered meticulously onto the yellow solution. After two days of slow diffusion, colorless fine square crystals of $[Fe(2-pytrz)_2\{Pt(CN)_4\}]\cdot 3H_2O$ were formed [45].

3.2. Magnetic and Structural Characterizations

Magnetic measurements of [Fe(2-pytrz)₂{Pt(CN)₄}]·3H₂O performed in Ref. [45] and reported in Figure 1, show the cooperative character of this SCO material which leads to a thermally-induced first-order spin transition occurring around \sim 152 K on cooling and \sim 154 K on heating, accompanied with a thermal hysteresis of ca. 2 K width. As already discussed in Ref. [45], Figure 1a, shows $\chi_m T$

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values of 3.51 cm 3 ·K·mol $^{-1}$ compatible with a HS (S=2) state of Fe(II) at high temperature, while below 150 K, this value drops abruptly to ca. 2.0 cm 3 ·K·mol $^{-1}$, indicating an incomplete spin transition, with the presence of $\sim 56\%$ residual HS fraction at low temperature. The slight decrease of this fraction at very low temperature (i.e., below 30 K) is attributed to zero field splitting effects, induced by the magnetic anisotropy of the current hexacoordinated Fe(II) ions. In agreement with X-rays diffraction data measurements [45], it was demonstrated that in the low-temperature regime, the spin transition involves one out of two atoms, as shown in Figure 1b. Indeed, the distortion experienced by one of the iron sites (see Figure 1b), originating from an elastic frustration occurring inside the lattice, induces the trapping of Fe1 in the HS state, thus precluding its transition at low-temperature.

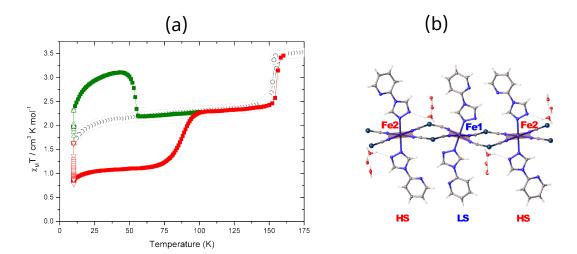


Figure 1. (a) Thermal behavior of the magnetic moment of the SCO compound $[Fe(2-pytrz)_2{Pt(CN)_4}]\cdot 3H_2O$, adapted from Ref. [45]. Black open circles represent the heating and cooling process without light showing the presence of a first order transition with thermal hysteresis around $T\sim 153$ K, between the HS and the intermediate HS-LS states. The green curve is the heating process after the green light ($\lambda=510$ nm) irradiation at the lowest temperature, and the red one corresponds to the heating process from the LS to the HS state through HS-LS state after the red light ($\lambda=830$ nm) irradiation at the lowest temperature. The relaxation temperatures, after green and red light excitations, are $T_{LIESST}\simeq 52$ K and $\simeq 90$ K, respectively. Temperature sweep rate is 0.4 K·min⁻¹ for all experiments. (b) Structure of $[Fe(2-pytrz)_2{Pt(CN)_4}]\cdot 3H_2O$ in the HS-LS phase.

In addition to the thermal hysteresis and the incomplete HS to HS-LS spin transition, Figure 1 reports the photomagnetic response of this compound under green light ($\lambda=510$ nm) showing a clear evidence of a LIESST (Light-Excited-Spin-State-Trapping) [45–48] effect allowing to access metastable photo-induced HS state at low-temperature. The resultant irradiated HS state relaxes back to the intermediate HS-LS state around ~52 K. On the other hand, the excitation of the intermediate HS-LS low-temperature phase using a red wavelength ($\lambda=830$ nm) produced the reverse-LIESST effect [45,47–49], leading to the fully "stable" LS state, whose thermal relaxation to the intermediate HS-LS phase takes place at $T\simeq90$ K. The investigations of the dynamical properties of this LS state allowed in a previous study [45] to demonstrate its stability at very low-temperature. It results that the intermediate HS-LS state is stable on cooling until some temperature located in the region 40–70 K and then becomes metastable due to its thermal freezing at lower temperatures.

This bistability between HS and HS-LS states was confirmed by DSC studies for which the thermal variation of the heat flow shows exo- and endothermic transitions at \sim 151.8 K and \sim 154.4 K, respectively (see Figure 2). The phase transition occurs with an enthalpy and entropy changes of $\Delta H = 4.8 \text{ kJ} \cdot \text{mol}^{-1}$ and $\Delta S = 30 \text{ J} \cdot \text{K}^{-1} \cdot \text{mol}^{-1}$, respectively, in agreement with the values reported in literature of SCO compounds [38,50]

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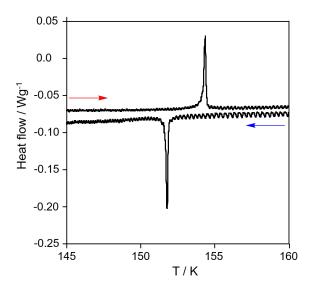


Figure 2. DSC study for $[Fe(2-pytrz)_2{Pt(CN)_4}] \cdot 3H_2O$ (scan rate = 0.3 K·min⁻¹) revealing the exo- and the endo-thermic transitions with maxima at 151.8 and 154.4 K, respectively. These values, derived from the maximum of the DSC peaks, are in good agreement with the transition temperatures obtained in magnetic data.

The change of the unit cell parameters as well as β angle misfit between HS and HS-LS states is depicted in Table 1. The latter is compared to two other SCO compounds that have been already studied by optical microscopy, namely [{Fe(NCSe)(py)₂}₂(m-bpypz)] [37,42–44] and [Fe(btr)₂(NCS)₂]·H₂O [35]. One can see that the change of β angle is much smaller for the title compound, which then exhibits reduced distortion at the transition.

Table 1. Relative variation of lattice parameters between the high- and low-temperature phases of several SCO compounds. The isotropic/anisotropic character of the volume change at the transition affects the HS-LS interface velocity and the resilience of the single crystal during the transition. HS, LS and HS-LS phases denote high spin, low-spin and intermediate HS-LS states, respectively.

Compound	Phases	$\frac{\Delta a}{a}$ (%)	$\frac{\Delta b}{b}$ (%)	$\frac{\Delta c}{c}$ (%)	$\Delta eta(^{\circ})$
$[{Fe(NCSe)(py)_2}_2(m-bpypz)]$	$HS \leftrightarrow LS$	0.6	1.58	0.3	-2.329
$[Fe(2-pytrz)_2{Pt(CN)_4}]\cdot 3H_2O$	$HS \leftrightarrow HS\text{-}LS$	1.82	2.17	0.69	-0.036
$[Fe(btr)_2(NCS)_2](H_2O)$	$HS \leftrightarrow LS$	-2.04	4.12	3.1	-1.22

To avoid the strong elastic effects and volume changes which usually degrade the materials, and to compare the crystal shape effects on the interface behavior (shape and velocity), we investigate the spatio-temporal features of the compound $[Fe(2-pytrz)_2\{Pt(CN)_4\}]\cdot 3H_2O$ for which the unit cell distortion between the low-temperature and the high-temperature phases is less significant than those of compounds $[\{Fe(NCSe)(py)_2\}_2(m-bpypz)]$ and $[Fe(btr)_2(NCS)_2](H_2O)$. Indeed in $[Fe(2-pytrz)_2\{Pt(CN)_4\}]\cdot 3H_2O$, the transition takes place between a fully HS phase at high temperature and an intermediate HS-LS phase at low-temperature, which then involves less volume change.

3.3. Optical Microscopy Measurements

Optical microscopy studies have been performed on a single crystal of the SCO compound [Fe(2-pytrz)₂{Pt(CN)₄}]·3H₂O [45] of size $\sim 400~\mu m~\times~300~\mu m$ which exhibits an incomplete transition between a fully HS state at high temperature and a well organized HS-LS phase at low temperature [45].

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3.3.1. Thermal Hysteresis of One Single Crystal

OM investigations evidencing the spatio-temporal properties of the present single crystal along its thermal spin transition are reported in Figure 3a, which displays a set of selected snapshots on cooling and heating processes, exhibiting different colors in the HS and HS-LS phases. Furthermore, a single domain nucleation and growth mechanism upon spin transition accompanied with a well-defined interface between the two phases, is observed. We exploited the thermochromic character of this transformation, which leads to which leads to different optical densities (OD) $[OD = \log_{10}(\frac{I_0}{I})]$, where I_0 is the incident bright field intensity and I is the transmitted intensity] for the HS and HS-LS states. The transition temperatures and the thermal hysteresis loop of the system, were evaluated using an image processing treatment, whose procedure was already explained in [38,51], the spatially averaged total OD emerging from the single crystal. The obtained thermal evolution of the OD is given in Figure 3b. It is worth mentioning that the local HS fraction, n(x,y,T), of the system connects linearly to the normalized local OD, defined as $n(x,y,T) = \frac{OD(x,y,T) - OD(HS - LS)}{OD(HS) - OD(HS - LS)}$ (where OD(HS) and OD(HS-LS) are the OD values in the HS and intermediate HS-LS states, respectively).

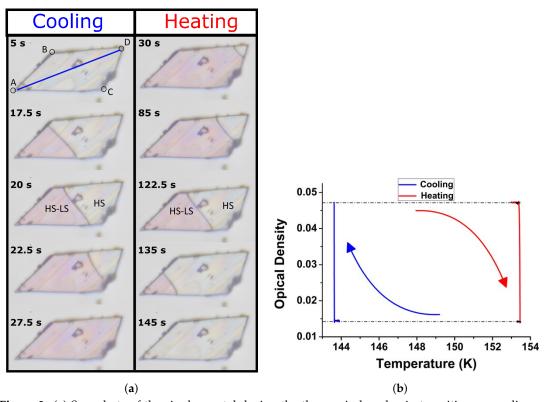


Figure 3. (a) Snapshots of the single crystal during the thermo-induced spin transition on cooling and heating. (b) Temperature dependence of the average OD showing the occurrence of a thermal hysteresis loop. The temperature sweep rate was 0.5 K.min⁻¹, for both cooling and heating processes.

The transition temperatures deduced from Figure 3b were found to be $T^- \sim 143$ K on cooling and $T^+ \sim 153$ K on heating. Compared to the magnetic data which led to $T^- \sim 152$ K and $T^+ \sim 154$, one may conclude to the existence of a net disagreement. However, one has to consider that the thermal hysteresis of OM data relates to the transition of a unique single crystal, while that of magnetic measurements (Figure 1a) resulted from the average response of a large number of crystalline powder (or micro-crystals) sample with distributed sizes, shapes, transition temperatures and thermal hysteresis.

OM images of Figure 3a, recorded during the spin transition indicate that the first-order transition, on cooling between the HS and the intermediate HS-LS phases, starts from the bottom left corner (point A) and proceeds as a unique domain that propagates until reaching the first upper right corner

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(point B). During this first stage, the interface has more or less, a circular shape with contact angles perpendicular to the crystal edges. From point B the two crystal borders become parallel, which affects the interface shape becomes sigmoidal (S-shaped). This special shape is imposed by: (i) the border conditions between the interface and the crystal borders (i.e., the contact angles) which remain perpendicular to each others and (ii) the shift existing along the crystal length between the downer and upper interface points. When the interface reaches point C, a similar behavior as that of the first stage (A \rightarrow B) occurs again, since its shape becomes again curved with contact angles equal to $\sim \frac{\pi}{2}$.

3.3.2. Interface Velocities

As we use the OM data to evaluate the thermal hysteresis loop of Figure 3b, we use it also to monitor the interface position along the diagonal line AD (see Figure 3a) as a function of time, the behavior of which is given in Figure 4. There, one can easily remark that the average velocity of the interface on cooling is higher than that on heating. We can also observe the existence of different propagation regimes with different velocities for both processes, attributed to the irregular shape of the single crystal, in good agreement with a similar study, recently realized on the analogous compound $[Fe(2-pytrz)_2Pd(CN)_4]\cdot 3H_2O$ [38]. Figure 4 also evidences the occurrence of an acceleration regime when the interface gets close to the extremity of the single crystal. It is interesting to notice that this result is in good agreement with the predictions of recent theoretical investigations [40,52]. The quantitative evaluation of the average propagation velocities in the first regime before the acceleration are $17 \, \mu m \cdot s^{-1}$ and $2.3 \, \mu m \cdot s^{-1}$ on cooling and heating processes, respectively.

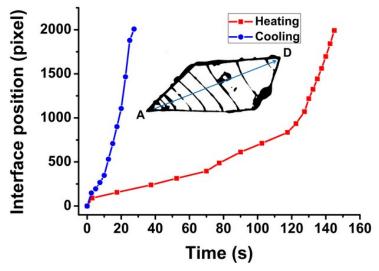


Figure 4. The interface position as a function of time, estimated along the diagonal of the crystal, from point A to point D and D to A on cooling and heating, respectively, indicated with blue and red symbols. Inset shows the interface positions on the crystal along the heating process.

In Figure 5, we present the time evolution of the interface position along the crystal edges upon cooling and heating. In Figure 5a, the interface velocity is evaluated along DB and DC directions during the heating process. The obtained curves display a very similar behavior with the presence of a regime change (an acceleration) starting around $t \simeq 80$ s for both of them. In the first regime on DB and DC directions, the measured interface velocities are $v_1 = 1.6~\mu\text{m}\cdot\text{s}^{-1}$ and $v_1 = 1.4~\mu\text{m}\cdot\text{s}^{-1}$, respectively, while for the second regime, we found the respective values, $v_2 = 3.7~\mu\text{m}\cdot\text{s}^{-1}$ and $v_2 = 3.1~\mu\text{m}\cdot\text{s}^{-1}$. The set of obtained results is summarized in Table 2. The existence of the above two regimes is here attributed to the observable crystal defect, visible in the inset of Figure 5a along DC direction, which perturbs the interface propagation process on heating. Figure 5b presents the time-dependence of

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the interface position on cooling along AB and AC directions. There also, we remark the existence of two regimes. The first one, in AB and AC directions is characterized by the interface velocities, $v_1 = 10 \ \mu \text{m} \cdot \text{s}^{-1}$ and $v_1 = 12 \ \mu \text{m} \cdot \text{s}^{-1}$, respectively, while the second regime has higher respective velocities, $v_2 = 22 \ \mu \text{m} \cdot \text{s}^{-1}$ and $v_2 = 32 \ \mu \text{m} \cdot \text{s}^{-1}$. Interestingly, both curves show a regime change at the same time, $t \simeq 8$ s, which was identified as a particular time at which OM images show the spontaneous emergence of an additional HS-LS phase ahead the front interface. To emphasize this point, we represent in Figure 6 the spatial dependence of the transmitted light intensity across the interface along the blue line reported in the snapshots of Figure 6. This investigation is performed for different positions of the interface during the cooling regime. The intensity profile across the interface (Figure 6) is characterized by the presence of a significant reduction of the light intensity signal in the interface region compared to those of HS and HS-LS phases. This decrease is due to the darker character of the front which results from the light diffusion caused by the HS/(HS-LS) interface. In the right panels of Figure 6, we represented the intensity profile along the blue lines associated to the images of the left panel. For the two regimes showed in (b) and (c) images, the intensity profile shows a single peak corresponding to a unique interface, although that of image (b) appears as little bit broader. In contrast, the intensity profile corresponding to image (a), associated to the particular time value, t = 8 s, for which the velocity regime change appeared in Figure 5b, shows the existence of two peaks. This indicates the presence at this particular time of a second interface emerging ahead the main front. The origin of this second interface is attributed to a nucleation starting from the bottom of the crystal and initiated by the presence of defects which accelerate the process of relaxation of the elastic strain accompanying the volume change at the transition.

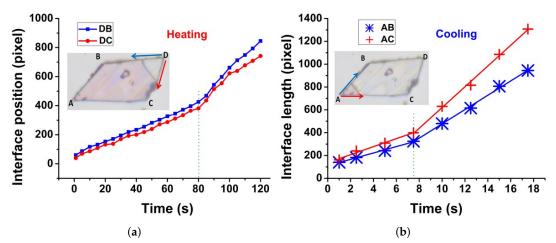


Figure 5. Interface velocity evaluated along the crystal edges as indicated in the two insets: (a) on heating along DB (blue) and DC (red) directions and (b) on cooling along AB (blue) and AC (red) directions. The vertical dashed lines indicate the times at which a regime change takes place in the propagation process.

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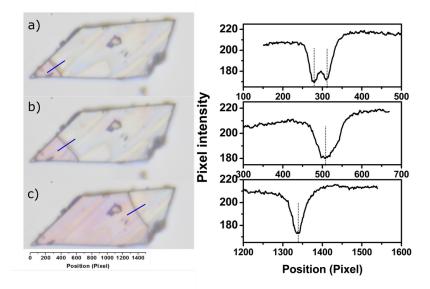


Figure 6. Intensity profile across the interface (separating HS and HS-LS phases) along the blue line during the cooling process. The pixel size is $0.3 \, \mu m \times 0.3 \, \mu m$.

Table 2. The various interface velocities evaluated on heating and cooling along AB, AC, DB and DC directions for the two different regimes identified in Figure 5.

Direction of Propagation		Velocity (μm·s ⁻¹)		
		1st Regime	2nd Regime	
Heating	DB	1.6	3.7	
	DC	1.4	3.1	
Cooling	AB AC	10 12	22 32	

OM allows also to derive relevant information about the change of the crystal dimensions during the spin transition process. Using our own image processing software developed on Matlab, we could measure the distances AB, AC and AD in the low-temperature and high temperature phases, from which we deduced their relative variations. The obtained results are summarized in Table 3. Thanks to the good agreement with the relative variation of the lattice parameters extracted from X-rays studies [38,45] and summarized in the same Table 3, one can easily identify that the front propagation takes place in the *a-b* plane, where the most important changes are observed at the transition.

Table 3. Relative variation of the crystal sizes derived from OM data and X-rays measurements [38,45] in the HS and HS-LS phases. The pixel size is \sim 0.3 \times 0.3 μ m².

X-rays Data							
Lattice Parameter (Å)	HS	HS-LS	Relative Variation (%)				
а	25.248	24.795	1.82				
b	7.4044	7.247	2.17				
С	27.293	27.105	0.69				
Optical microscopy data							
Distances (pixel)	HS	HS-LS	Relative variation (%)				
d_{B-D}	1209	1197	1				
d_{A-B}	950	929	2.2				
d_{A-C}	1542	1522	1.2				

4. Theoretical Section

4.1. The Hamiltonian

The present experimental results are modeled using the kinetic version of an Ising model [53] in which the low- and the high-temperature phases are assumed to be LS and HS, respectively. We demonstrated in a previous work, that the spatiotemporal version of such a description leads to a reaction diffusion equation [38,54] that governs the time and space dependence of the HS fraction during the spin transition. The Hamiltonian of the Ising-like model is expressed as follows,

$$\mathscr{H} = -J\sum_{ij} S_i S_j + \Delta_{eff} \sum_i S_i. \tag{1}$$

In Equation (1), S_i is a fictitious operator whose eigenvalues +1 and -1 are respectively associated with the HS and LS states and J>0 stands for the ferro-elastic interactions between the spin states, while $\Delta_{eff}=\Delta-\frac{k_BT}{2}\ln g$ is the effective energy gap, which contains the contributions of the ligand-field energy, Δ , and the degeneracy ratio, $g=\frac{g_{HS}}{g_{LS}}$, between the HS and LS states. The latter contribution enters under the form of an entropic term in the effective ligand-field energy. The mean-field analysis of Hamiltonian (1) leads quite easily to the following expression of the free energy,

$$\mathcal{F}^{hom} = \frac{1}{2} Jm^2 - k_B T \ln \left[2g \cosh\left(\frac{Jm - \Delta_{eff}}{k_B T}\right) \right], \tag{2}$$

where $m = \langle s \rangle$, is the average fictitious magnetization per site and T is the temperature. This net magnetization connects to the HS fraction, n_{HS} , through the simple relation

$$n_{HS} = \frac{1+m}{2} \tag{3}$$

The used parameters for this simulation are $\Delta=380$ K, $\ln g=5$ and J=300 K. The transition temperature obtained with the model is here $T_{\frac{1}{2}}=\frac{2\Delta}{k_B\ln(g)}\simeq 150$ K. The current model does not account for the elastic effects between the SCO sites nor for the volume change at the transition between LS and HS states. The equilibrium properties of the mean field version of Hamiltonian (1) have been carefully studied in the literature; they are obtained by solving the equation, $\frac{\partial \mathcal{F}^{hom}}{\partial m}=0$, which gives the state equation,

$$m = \tanh \beta (Jm - \Delta_{eff}),$$
 (4)

whose resolution leads to specify the conditions of obtaining first-order and gradual spin transition [38,54]. This model is then quite well adapted to mimic the spatio-temporal behavior of SCO materials showing first-order transitions with low volume change. Here, we are mainly interested in the correlation between the dynamics of the HS-LS interface (as well as its shape along the propagation process) and its relation with the global macroscopic crystal's shape. Therefore, to be as realistic as possible, we designed in the simulation a lattice which has a shape resembling that of the current crystal, shown in Figure 3.

4.2. Spatio-Temporal Aspects of the HS Fraction

According to the experimental results of OM, summarized in Figure 3, the interface propagates along the diagonal direction of the lattice, with different velocities along the lattice edges (Figure 4). This behavior supports the idea of the existence of strong anisotropic effects in the propagation process. To include this important property, we extended our previous isotropic reaction diffusion model [54], for which we give the main developments below.

Derivation of the Anisotropic Reaction Diffusion Approach

First, we start with the homogeneous mean-field free energy of Equation (2), that we expanded as function of $n_{HS}(\vec{r},t)$ and $\vec{\nabla} n_{HS}$, where \vec{r} is the position and t, the time. The density of free energy at position \vec{r} is then written as $f(\vec{r}) = f(n_{HS}, \vec{\nabla} n_{HS})$ for which corresponds the density of the homogeneous free energy $f^{hom}(n_{HS}) = f(n_{HS}, \vec{\nabla} n_{HS})$ which relates to \mathcal{F}^{hom} through $\mathcal{F}^{hom} = V \times f^{hom}(n_{HS})$, where V denotes the system's volume (here, the surface in 2D). Following a standard procedure, $f(n_{HS}, \vec{\nabla} n_{HS})$ is expanded as a function of successive powers of the gradient of the HS fraction, giving

$$f(n_{HS}, \vec{\nabla}n_{HS}) = f^{hom}(n_{HS}, 0) + \vec{L} \cdot \vec{\nabla}n_{HS} + \frac{1}{2} \vec{\nabla}^T \cdot D_{ij} \cdot \vec{\nabla}n_{HS}, \tag{5}$$

where,

$$\vec{L} = [L_{x_1}, L_{x_2}, L_{x_3}] = \frac{\partial f}{\partial (\partial n_{HS}/\partial x_i)},\tag{6}$$

is a vector evaluated at zero gradient, and D_{ij} is a tensor property known as the gradient-energy coefficient with components,

$$D_{ij} = \frac{1}{2} \frac{\partial^2 f}{\partial (\partial n_{HS} / \partial x_i) \partial (\partial n_{HS} / \partial x_i)}, \tag{7}$$

which should be a symmetric tensor. Since the free energy density does not depend on the direction of the gradients, we take L=0. Furthermore, if the homogeneous system has an inversion center, \vec{L} is automatically zero, and if in addition the system is isotropic or cubic, D_{ij} will be a diagonal tensor with equal components. In Equation (5), $\vec{\nabla}^T$ is the transpose of the gradient operator. So, in the general case of an anisotropic material, the density of free energy is given by,

$$f(n_{HS}, \vec{\nabla}n_{HS}) = f^{hom}(n_{HS}, 0) + \frac{1}{2}\vec{\nabla}^T \cdot D_{ij} \cdot \vec{\nabla}n_{HS}.$$
 (8)

and the equation of motion of the HS fraction, including the spatial variations, is expressed as,

$$\frac{\partial n_{HS}}{\partial t} = -\frac{\partial f}{\partial n_{HS}}. (9)$$

The local diffusion potential $f(\vec{r})$ at instant $t = t_0$ can be determined from the variation of the rate of the total free energy, \mathcal{F} , as follows

$$\mathcal{F}(t_0) = \int_V \left[f^{hom} \left(n_{HS}(\vec{r}, t_0) \right) + \vec{L} \cdot \vec{\nabla} n_{HS} + \frac{1}{2} \vec{\nabla}^T n_{HS} D_{ij} . \vec{\nabla} n_{HS} \right] dV, \tag{10}$$

where, we have kept the general expression of $f(\vec{r})$, given in Equation (5). Considering that the order parameter varies with the local velocity, as $n_{HS}(\vec{r},t) = n_{HS}(\vec{r},t_0) + n_{HS} \times t$, we can estimate the rate of variation of \mathcal{F} at t_0 , as follows

$$\frac{d\mathcal{F}}{dt}\Big|_{t_0} = \int_V \left[\frac{\partial f^{hom} \left(n_{HS}(\vec{r}, t_0) \right)}{\partial n_{HS}} n_{HS}^{\cdot} + \vec{L} \cdot \vec{\nabla} n_{HS}^{\cdot} + \vec{\nabla}^T n_{HS}^{\cdot} D_{ij} \cdot \vec{\nabla} n_{HS} \right] dV. \tag{11}$$

Using the general identity

$$\vec{\nabla}^T \left(\dot{p} D_{ij} \vec{\nabla} p \right) = \dot{p} \vec{\nabla}^T D_{ij} \vec{\nabla} p + \vec{\nabla}^T \dot{p} D_{ij} \cdot \vec{\nabla} p, \tag{12}$$

which is combined to the divergence theorem, and applying Newman border conditions which zeros the gradient of the HS fraction at the lattice borders one easily finds the following expression for the rate of variation of \mathcal{F} ,

$$\frac{d\mathcal{F}}{dt}\Big|_{t_0} = \int_V \left[\frac{\partial f^{hom} \left(n_{HS}(\vec{r}, t_0) \right)}{\partial n_{HS}} - \vec{\nabla}^T D_{ij} \cdot \vec{\nabla} n_{HS} \right] n_{HS} dV. \tag{13}$$

Since the lattice parameters varies with a quantity $\delta n_{HS} = \dot{n}_{HS} \times \delta t$, the change in the total potential, $\delta \mathcal{F}$ is then the sum of local variations and can be written as,

$$\delta \mathcal{F} = \int_{V} \left[\frac{\partial f^{hom} \left(n_{HS}(\vec{r}, t_{0}) \right)}{\partial n_{HS}} - \vec{\nabla}^{T} D_{ij} \cdot \vec{\nabla} n_{HS} \right] \delta n_{HS} \left(\vec{r} \right) dV. \tag{14}$$

The quantity between brackets in Equation (14) is the change of the density of local free energy resulting from the change of HS fraction. The equation of motion is then simply given by

$$\frac{\partial n_{HS}}{\partial t} = -\frac{\delta \left(\frac{d\mathcal{F}}{dV}\right)}{\delta n_{HS}}.$$
 (15)

4.3. The Spatio-Temporal Equation of Motion of the HS Fraction

The general spatio-temporal equation of motion of the HS fraction for an anisotropic diffusion of the spin states, is deduced from Equation (15) as follows,

$$\frac{\partial n_{HS}}{\partial t} = -\Gamma \frac{\partial \mathcal{F}^{hom}}{\partial n_{HS}} + \vec{\nabla}^T D_{ij} \cdot \vec{\nabla} n_{HS}, \tag{16}$$

where Γ is a time scale factor. The anisotropy of the diffusion constant may have several origins, among which we quote (i) the anisotropic deformation of the unit cell at the transition between the LS and HS states, as already demonstrated for the SCO compound [{Fe(NCSe)(py)₂}₂(m-bpypz)] [37], or (ii) the existence of direction-dependent elastic constants acting between the SCO units, as well as (iii) the presence of a crystalline misorientation (a jump in angle across the boundary) which leads to an inclination angle of the boundary plane between the LS and the HS phases. On the other hand, ideally the gradient ∇n_{HS} and the diffusion tensor D_{ij} (given in Equation (7)) should be calculated in the crystalline frame, which is not the same in the intermediate HS-LS and HS states of the title compound.

Because of the change of the unit cell structure at the transition, the direct relation between the tensor D_{ij} and crystalline frame is far from obvious. However, if one considers that the crystal structure remains unchanged, and the misorientation angle at the interface is zero, then the diffusion tensor can be written in the general form as $D_{ij} = R_{\theta}^T D_{ij}^{diag} R_{\theta}$, where D_{ij}^{diag} is the diagonal diffusion constant, and R_{θ} is the transfer matrix where the superscript "T" stands here for the transpose. The matrices D_{ij}^{diag} and R_{θ} have the following expressions,

$$D_{ij}^{diag} = \begin{pmatrix} D_1 & 0 \\ 0 & D_2 \end{pmatrix} \quad \text{and} \quad R_{\theta} = \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix}. \tag{17}$$

By developing D_{ij} as function of D_1 , D_2 and θ , whose expressions are introduced in Equation (16), the following general equation of motion is obtained:

$$\frac{\partial n_{HS}}{\partial t} = -\Gamma_n \frac{\partial \mathcal{F}^{hom}}{\partial n_{HS}} + \left(D_{xx} \frac{\partial^2}{\partial x^2} + D_{yy} \frac{\partial^2}{\partial y^2} + D_{xy} \frac{\partial^2}{\partial x \partial y} \right) n_{HS}, \tag{18}$$

where, the expressions of the three components of the diffusion tensor, connect to D_1 , D_2 and θ , as follows,

$$D_{xx} = D_1 \cos^2 \theta + D_2 \sin^2 \theta,$$

$$D_{yy} = D_1 \sin^2 \theta + D_2 \cos^2 \theta,$$

$$D_{xy} = (D_1 - D_2) \cos \theta \sin \theta.$$
(19)

According to Equation (7), the elements of the diffusion tensor, D_{ij} , have the following expressions,

$$D_{xx} = \frac{1}{2} \frac{\partial^2 f}{\partial^2 (\partial n_{HS}/\partial x)},$$

$$D_{yy} = \frac{1}{2} \frac{\partial^2 f}{\partial^2 (\partial n_{HS}/\partial y)}$$

$$D_{xy} = \frac{1}{2} \frac{\partial^2 f}{\partial (\partial n_{HS}/\partial x) \partial (\partial n_{HS}/\partial y)}.$$
(20)

Equation (18) expresses the motion of a front interface in the case of an anisotropic diffusion of spin states. In these qualitative simulations, we used $D_1 = 2.0 \, \mu \text{m}^2 \cdot \text{s}^{-1}$, $D_2 = 15.0 \, \mu \text{m}^2 \cdot \text{s}^{-1}$ and $\theta = 30^\circ$, which were found to well reproduce the experimental shapes of the interface. However, it should be noted that other triplets of solutions may also exist.

It is interesting to discuss briefly some limiting cases. First of all, in the stationary and uniform state, i.e., $\frac{\partial n_{HS}}{\partial t}=0$ and and $\nabla^2 n_{HS}=0$, Equation (18) gives the self-consistent mean-field Equation (4), which describes the equilibrium properties of the SCO material. Furthermore, one can easily see that in the case of an isotropic diffusion of the spin states, i.e., $D_1=D_2=D$, Equation (18) leads to $D_{xy}=0$ and $D_{xx}=D_{xy}=D$, which allows to recover the previous equation of motion,

$$\frac{\partial n_{HS}}{\partial t} = -\Gamma_n \frac{\partial \mathcal{F}^{hom}}{\partial n_{HS}} + D\Delta n_{HS},\tag{21}$$

that we have established in Ref. [54] to describe the front propagation observed in the SCO compound [{Fe(NCSe)(py)₂}₂(m-bpypz)]. The solutions of Equation (18) have the form (in 1D) $n_{HS}(x,t) = a + b \tanh(\frac{x-vt}{\delta})$ where v is the interface velocity and δ is its width. Here v is proportional to \sqrt{D} , and the propagation direction takes place along the crystal length with a right interface as we demonstrated in a previous report [54], which does no meet the experimental results of Figure 3.

Unfortunately, the analytical resolution of the anisotropic reaction diffusion Equation (18) is out of reach, although one may expect that the existence of several components in the Laplacian of the HS fraction will drive the front propagation along a direction depending on the ratio $\frac{D_2}{D_1}$.

Equation (18) is then solved numerically, after discretization of time and space using finite difference method, where the used space (dx, dy) and time (dt) steps values are, dx = dy = 0.2 µm and dt = 0.001 s.

As clearly shown by Figure 7, the calculated snapshots are in excellent agreement with the experimental results proving the reliability of the anisotropic reaction diffusion model in the description of the spatio-temporal features of the front propagation accompanying the first-order transitions of $[\text{Fe}(2\text{-pytrz})_2\{\text{Pt}(\text{CN})_4\}]\cdot 3\text{H}_2\text{O SCO}$ single crystal. In particular, it is found that the anisotropy of the spin states diffusion tensor, D_{ij} , allows here to control the direction of propagation of the interface, which depends on the angle θ and the ratio $\frac{D_2}{D_1}$. Furthermore, the shape of the simulated HS/LS interface, its variation along the transformation process and the obtained contact angles with the "crystal" edges are very similar to the experimental ones (see Figure 7).

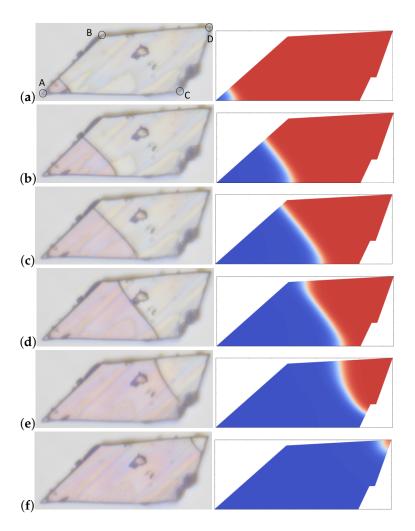


Figure 7. (a–f) Left panels present the experimental optical microscopy images of the crystal transformation of the compound $[Fe(2-pytrz)_2{Pt(CN)_4}]\cdot 3H_2O$ on cooling along the thermal hysteresis of Figure 3b. Right panels present the simulated snapshots obtained from the resolution of the anisotropic Equation (18), showing the spatiotemporal configurations of the lattice along the nucleation and growth of the LS phase (in blue) inside the HS (in red) phase on cooling. The parameter values are $D_1 = 2.0 \ \mu m^2 \cdot s^{-1}$, $D_2 = 15.0 \ \mu m^2 \cdot s^{-1}$ and $\theta = 30^\circ$ and the temperature is 150 K.

Furthermore, it is an interesting extension to consider the direct connection between the tensor elements D_1 , D_2 and θ or D_{xx} , D_{xy} , D_{yy} and the crystallographic data. For that the gradients of HS fraction, should be expressed in the crystallographic basis of the material. Let us denote by \vec{a} and \vec{b} the two vectors of unit cell attached to the material in which a point M has the coordinates (x',y') and by \vec{i} , \vec{j} the orthonormal system attached to the laboratory framework, where the same point has the coordinates (x,y), so that we have $OM = x\vec{i} + y\vec{j} = x'\vec{a} + y'\vec{b}$. The transfer matrix between the two basis, denoted P, is made of scalar products between the two vectors of the sets of basis. It is easy to demonstrate that gradient operator in the crystal frame is expressed,

$$\vec{\nabla}' = A \frac{\partial}{\partial x} \vec{a} + B \frac{\partial}{\partial y} \vec{b}, \tag{22}$$

where $A = \frac{1}{\frac{\partial x'}{\partial x} + \frac{\partial y'}{\partial x}}$ and $B = \frac{1}{\frac{\partial x'}{\partial y} + \frac{\partial y'}{\partial y}}$ depend on the scalar products between the crystal and laboratory set of basis defined above.

Since the objective of this part is to re-write the equation of motion (18) of the HS fraction in the crystal frame, it is enough to focus on the diffusion part, i.e., $\vec{\nabla}'.D'.\vec{\nabla}'$, whose developed expression can be written,

$$\vec{\nabla}'.D'_{ij}.\vec{\nabla}' = A^2 a^2 D'_{x'x'} \frac{\partial^2}{\partial x^2} + B^2 b^2 D'_{y'y'} \frac{\partial^2}{\partial y^2} + A \vec{a}.\vec{b} \left(D'_{x'y'} + D'_{y'x'} \right) \frac{\partial^2}{\partial x \partial y}$$
(23)

The comparison between Equations (18) and (23) in which the diffusion parts should be equal, allows to connect the tensor elements of the diffusion tensors D' and D, as follows:

$$D_{xx} = A^{2}a^{2}D'_{x'x'},$$

$$D_{yy} = B^{2}b^{2}D'_{y'y'},$$

$$D_{xy} + D_{yx} = A \times B\vec{a}.\vec{b} \left(D'_{x'y'} + D'_{y'x'}\right).$$
(24)

These equations help to better understand how the lattice structure affects the diffusion tensor and so the direction of propagation of the HS/LS interface. These interesting features will be examined in a further work.

5. Conclusions

We have presented the spatio-temporal aspects of the spin transition derived from OM experiments, of a new robust single crystal of the compound [Fe(2-pytrz)₂{Pt(CN)₄}]·3H₂O showing a clear correlation between the HS/HS-LS interface properties at the thermal transition and the macroscopic crystal shape. In particular, the shape, the length and the velocity of the interface adapts to the change of the macroscopic crystal shape during the propagation process. This interplay between the electro-elastic HS/HS-LS interface and the macroscopic characteristics of the crystal validates our belief that the origin of the elastic interactions in SCO materials are of long-range nature, as predicted in related elastic models [37,39–42]. In addition, the analysis of the total crystal length change derived from OM data enabled to find an excellent agreement with X-rays data [45], thus enhancing the relevance of this technique. Finally, the present experimental spatio-temporal findings are adequately described by a reaction diffusion formalism which allowed to reproduce the main experimental features, such as (i) the diagonal character of the front propagation and (ii) the change of the interface shape according to that of the crystal during the transition. To reproduce these aspects, we extended our previous spatio-temporal mean-field dynamics to include anisotropic effects of the spin state "diffusion", resulting in a more general description, which can be studied for its own. This work highlights the importance of synthesis of cooperative SCO materials showing incomplete transitions, which present less volume change, thus enhancing their resilience at the transition. Experiments on light driven interface control are in progress to explore the dynamical response of the interface to such excitations.

Author Contributions: H.F. did the optical microscopy measurements and made the image processing of the data. G.B. helped in the technical developments of the experimental setup of cryogenic optical microscopy. M.P.-E. worked on the experimental modeling of the spatiotemporal data and did the simulations of the reaction diffusion model. S.T., synthesized the sample, did and analyzed the crystallographic and the calorimetric measurements. K.B. supervised the optical microscopy and theoretical works and wrote the paper with the contribution of all authors.

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